Applicant: Thomas J. McMurry, et al.

Attorney's Docket No.: 13498-005004 / MET-4

Serial No.: 10/755,506 Filed: January 12, 2004

Page : 2 of 17

Amendments to the Specification:

Please amend the paragraph after the title on page 1 as follows:

CROSS-REFERENCE TO RELATED APPLICATIONS

This application is a continuation (and claims the benefit of priority under 35 USC § 120) of United States Patent Application No. 10/034,522, filed December 20, 2001, now U.S. Pat. No. 6,676,929, which is a continuation of United States Patent Application No. 08/875,365, filed December 12, 1997, which is a §371 application of International Patent Application No. PCT/US96/00164 (WO 96/23526), filed January 16, 1996, which is a continuation-in-part of United States Patent Application 08/382,317, filed February 1, 1995, all of which are incorporated by reference in their entirety herein. This case is also related to U.S. Ser. No. 10/354,723, filed January 30, 2003 and U.S. Ser. No. 10/755,507, filed January 12, 2004.

Please replace the paragraph beginning at page 18, line 20 with the following paragraph:

Examples of chemical groups which would serve as a BHEM include carbon,
phosphorous, tungsten, molybdenum, or sulfur atoms having attached charged or neutral
heteroatoms such as oxygen, nitrogen, sulfur or halogens (especially fluorine) possessing two or
more lone electron pairs (i.e., full or partial negative charge) or electropositive hydrogen atoms
(i.e., protonated amine) for hydrogen bonding with water. These include groups such as sulfone,
ether, urea, thio-urea, amine, sulfonamide, carbamate, peptide, ester, carbonate and acetals.
Preferred groups include those which possess one or more partial or full negative charges in
aqueous solution at physiological pH wherein the negatively charged atoms cannot be partially
or fully neutralized by covalent or coordinate covalent bonding to the IEM. Examples of these
preferred BHEMs include negatively charged groups such as phosphate mono-ester, phosphate
diester, carboxylate, and sulphonate. More preferred are those which have phosphate groups or
any ester forms thereof. Even more preferred are phosphate diesters, since: a) they are highly
hydrophilic with four hydrogen-bonding oxygens; b) they are relatively readily synthesized using
techniques shown below; c) they serve as excellent linkers between the IEM and the PPBM; and

Applicant: Thomas J. McMurry, et al. Attorney's Docket No.: 13498-005004 / MET-4

Serial No.: 10/755,506 Filed: January 12, 2004

Page : 3 of 17

d) because phosphate compounds exist and are metabolized naturally in the body, phosphate diester-containing contrast agents are expected to be non-toxic.

Please replace the paragraph beginning at page 28, line 14 with the following paragraph:

If the moieties of this invention are positioned in the contrast agent as in structure (1)

above, the BHEM is preferably sulfone, urea, thio-urea, amine, sulfonamide, carbamate, peptide, ester, carbonate, acetals and more preferably

$$Y^3$$
 X^4 Y^4 Y_2 Y_2 Y_2 or ester forms.

where
$$Z = P$$
, W, Mo, or S
 Y^1 , $Y^2 = O$ or S
 Y^3 , $Y^4 = O$, S or not present
 $R_2' = H$, C_{1-6} alkyl or not present.

Most preferably, the BHEM is a phosphate group.

Please replace the paragraph beginning at page 28, line 37 with the following paragraph:

If the moieties of this invention are positioned in the contrast agent as in structure (2)

above, the BHEM is preferably sulfone, urea, thio-urea, amine, sulfonamide, carbamate, peptide,

ester, carbonate, acetals and more preferably the BHEM has the following formula:

$$Y^3$$
 X^4 Y^2 Y^2 Y^2 or ester forms,

Applicant: Thomas J. McMurry, et al.

Attorney's Docket No.: 13498-005004 / MET-4

Serial No.: 10/755,506 Filed: January 12, 2004

Page : 4 of 17

where
$$Z = P$$
, W, or Mo
 Y^1 , $Y^2 = O$ or S
 Y^3 , $Y^4 = O$, S or not present
 $R_2' = H$, C_{1-6} alkyl or not present.

Most preferably, the BHEM is a phosphate group.

Please replace the paragraph beginning at page 29, line 19 with the following paragraph: If the moieties of this invention are positioned in the contrast agent as in structure (3)

above, the BHEM is preferably SO₃ or ester forms, sulfone, urea, thio-urea, amine, sulfonamide, carbamate, peptide, ester, carbonate, acetal and more preferably

$$Y^3$$
 X^3 Y^4 Y_2 Y_2 Y_2 or ester forms,

where
$$Z = P$$
, W , Mo , or S

$$Y^{1}, Y^{2} = O \text{ or } S$$

$$Y^{3}, Y^{4} = O, S \text{ or not present}$$

$$R_{2}' = H, C_{1-6} \text{ alkyl or not present}.$$

Most preferably, the BHEM is a phosphate group.

Please replace the paragraph beginning at page 31, line 14 with the following paragraph:

For contrast agents comprising the formulas shown above, the metal ion M is more preferably Gd(III), Fe(III), Mn(II), Mn(III), Cr(III), Cu(III), Dy(III), Tb(III), Ho(III), Er(III) or Eu(III), and most preferably Gd(III). The BHEM is preferably sulfone, ether, urea, thio-urea,

Applicant: Thomas J. McMurry, et al. Attorney's Docket No.: 13498-005004 / MET-4

Serial No.: 10/755,506 Filed: January 12, 2004

Page : 5 of 17

amine, amide, sulfonamide, carbamate, peptide, ester, carbonate, acetal and more preferably COO or ester forms, SO₃ or ester forms and

$$Y^3$$
 X^4 Y^2 Y^4 Y^4 Y^4 or ester forms.

where
$$Z = P$$
, W, Mo, or S
 Y^1 , $Y^2 = O$ or S
 Y^3 , $Y^4 = O$, S or not present
 $R_2' = H$, C_{1-6} alkyl or not present.